



**CHEG 2334 Z01 & Z81**  
**Chemical Engineering Thermodynamics I**  
**Summer 2024**

Course Information	Description
<b>Instructor:</b>	Dr. Carl McIntyre, Adjunct Professor of Chemical Engineering
<b>Section # and CRN:</b>	Section Z01 (CRN 30718)
<b>Office Location:</b>	Wilson 201A
<b>Office, Mobile Phones:</b>	936-261-9406 (Office), 281-217-1169 (Mobile)
<b>Email Address:</b>	<a href="mailto:ecmcintyre@pvamu.edu">ecmcintyre@pvamu.edu</a>
<b>Office Hours:</b>	M-R 1000-1050
<b>Mode of Instruction:</b>	ONLINE ASYNCHRONOUS
<b>Course Location:</b>	To be determined (check your schedule in <i>Panthertracks</i> or at link below) <a href="https://pv-ssbw-201.pvamu.edu:9020/PROD/bwckschd.p_disp_dyn_sched">https://pv-ssbw-201.pvamu.edu:9020/PROD/bwckschd.p_disp_dyn_sched</a>
<b>Class Days &amp; Times:</b>	ASYNCHRONOUS INSTRUCTION
<b>Catalog Description:</b>	CHEG 2334 Chemical Engineering Thermodynamics I: 3 semester hours credit (3 lecture contact hours plus 1 lab contact hour). Introduction to chemical engineering calculations. PVT properties of fluids, equations of state. First and second laws of thermodynamics. Applications to heat effects and flow processes.
<b>Prerequisites:</b>	Prerequisites: <a href="#">CHEG 2333</a> or CHEG 2053.
<b>Co-requisites:</b>	None.
<b>Required Text(s):</b>	Dahm, K.D. and Visco, D.P., 2014. Fundamentals of chemical engineering thermodynamics. Nelson Education. A brand new book, FUNDAMENTALS OF CHEMICAL ENGINEERING THERMODYNAMICS makes the abstract subject of chemical engineering thermodynamics more accessible to undergraduate students.
<b>Other Resources:</b>	(1) LearnChemE.com (last accessed 1/5/2022) will be a source of video lectures <a href="http://www.learncheme.com/">http://www.learncheme.com/</a> . In particular, many screencasts on thermodynamic concepts and applications, may be found at the link below. <a href="http://www.learncheme.com/screencasts/thermodynamics">http://www.learncheme.com/screencasts/thermodynamics</a> . (2) Devoe (last accessed 1/18/2021) is author of an open source textbook <a href="http://www2.chem.umd.edu/thermobook/">http://www2.chem.umd.edu/thermobook/</a> .

**Course Learning Objectives:**

	Upon successful completion of this course, students will be able to accomplish these objectives	Student Learning Outcome # Alignment	Core Curriculum Objective Alignment
1	Demonstrate knowledge of the 1 <sup>st</sup> , 2 <sup>nd</sup> and 3 <sup>rd</sup> laws of thermodynamics	1, 2, 3, 4	N/A
2	Apply knowledge of 1 <sup>st</sup> and 2 <sup>nd</sup> laws of thermodynamics to identify, formulate and solve problems in energy conservation, power cycles, refrigeration cycles and liquefaction systems	1, 2, 3, 4	N/A
3	Demonstrate use of modern tools of the profession in solving basic material and energy flow problems in thermodynamics	1, 2, 3, 4	N/A

## Major Course Requirements

This course will utilize the following instruments to determine student grades and proficiency of the learning outcomes for the course. The course has been designed to ensure that students acquire a solid grounding in identifying, formulating and solving fundamental problems in thermodynamics.

## Method of Determining Final Course Grade

Course Grade Requirement [Name each major requirement]		Value	Total
1)	Discussion/Forum Activity	+10%	10
2)	Quiz, Notes, Exercise	+15%	15
3)	Video Presentation	+25%	25
4)	Project Report or Exam	+50%	50
5)	Non-Participation Discount (Optional)*	-10%	10
<b>Total:</b>		<b>+100%*</b>	<b>100*</b>

\*Less discount, if applicable

## Grading Criteria and Conversion:

A = 90 - 100

B = 80 - 89.9

C = 70 - 79.9

D = 60 - 69.9

F = Below 60

If a student has stopped attending the course (i.e. "stopped out") at any point after the first day of class but did not officially withdraw from the course and has missed assignments and exams, including the final exam, and performed below the grade level of a D, a grade of FN (failed-nonattendance) will be assigned for the final course grade to ensure compliance with the federal Title IV financial aid regulations. In contrast, if the student has completed all assignments and exams, including the final exam, but performed below the grade level of a D, a grade of F will be assigned for the final course grade.

## Detailed Description of Major Assignments

Assignment Title or Grade Requirement	Description
Discussion Grades (10%)	Topics for discussion will be posted by the instructor each week, to which students are expected to post a comment at the forum and also a response to another student's post at that same forum.*
Quiz Grades (15%)	Reading assignments will be given, for which each student is required to take notes** for upload to complete the assignment. Likewise, screencasts (video lectures) will be assigned, for which students will likewise take notes for later upload. Exercises (similar to the homework concept) will be assigned for completion and submission by upload to eCourses. Quizzes will be administered both in class and in eCourses.
Video Upload Grades (25%)	The video upload assignments will be individual or team based, as identified by instructor in each assignment. Each video assignment will require each student to be recorded (seen and heard) explaining and/or demonstrating a concept that the student has learned via assigned readings, screencasts (video lectures), class meetings, individual or group study, etc. Typical length will be 5-8 minutes.
Exam Grades (50%)	Students are expected to be present (face-to-face) for exams, of which there will be a minimum of 2, one in October during the scheduled midterm examination period and one in November just before the Thanksgiving break. The location of the exams will be announced (these will take place in rooms large enough for social distancing). Masks will be required to be allowed in the exam room.
Project Grades (50%)	The projects will be individual or team based, as identified by instructor in each assignment. Each project will require open ended problem solution, in some cases using computing tools such as Aspen or HYSYS simulation software, accessible via VDI. The results are to be communicated in a letter report, with style and content requirements as communicated by the instructor.

\* The first student to post will have to return to the forum later, after another student has also posted, in order to post a comment to another student's post.

\*\* The style of note taking is the Bullet Point Notes method, used in the Guaranteed 4.0 Plan, for which guidance will be posted at the eCourses site.

## **Course Procedures or Additional Instructor Policies**

### **Taskstream**

Taskstream is a tool that Prairie View A&M University uses for assessment purposes. One of your assignments may be required to be submitted as an "artifact," an item of coursework that serves as evidence that course objectives are met. If applicable, more information will be provided during the semester, but for general information, you can visit Taskstream via the link in eCourses.

### Semester Calendar\*

Week	Lecture Topic	Assignments
1	Review of syllabus; thermo. background, dimensions and units, size and amounts, force, temperature, pressure, work, energy, and heat (Ch. 1) PL1	<ul style="list-style-type: none"> <li>• Upload BPN for Ch. 1</li> <li>• Upload BPN for PL1**</li> <li>• Complete Quiz Ch. 1</li> <li>• Upload Exercise Ch. 1</li> </ul>
2	Joule's experiments, internal energy, 1 <sup>st</sup> law of thermodynamics, energy balance of close systems, state and path functions; (Ch. 2). PL2A	<ul style="list-style-type: none"> <li>• Upload BPN: §2.1-2.5</li> <li>• Online Quiz §2.1-2.5</li> <li>• Upload Exercise §2.1-2.5</li> </ul>
3	Equilibrium, phase rule, DOF, reversible, isochoric, isobaric processes, Enthalpy; heat capacity, mass and energy balances for open systems (Ch. 2) PL2B	<ul style="list-style-type: none"> <li>• Upload BPN: §2.6-2.10</li> <li>• Online Quiz §2.6-2.10</li> <li>• Upload Exercise §2.6-2.9</li> </ul>
4	PVT behavior of pure substances, PV diagram, single phase region, Virial equation of state, ideal gas law; equations for process calculations: isothermal, isobaric, isochoric, and adiabatic processes (Ch. 3) PL3A	<ul style="list-style-type: none"> <li>• Upload BPN: §3.1-3.5</li> <li>• Online Quiz §3.1-3.5</li> <li>• Upload Exercise §3.1-3.5</li> </ul>
5	Polytropic processes, Irreversible processes; application of virial eqns; cubic equations of state, generalized correlations for gases & liquids (Ch. 3); PL3B	<ul style="list-style-type: none"> <li>• Upload BPN: §3.6-3.9</li> <li>• Online Quiz §3.6-3.9</li> <li>• Upload Exercise §3.6-3.8</li> </ul>
6	Heat effects; sensible heat effects, T-dependence of heat capacity, latent heats of pure substances, (Ch. 4) <b>Project 1 First Law Analysis of Power Cycle Components</b>	<ul style="list-style-type: none"> <li>• Upload BPN: §4.1-4.4</li> <li>• Online Quiz §4.1-4.4</li> <li>• Upload Exercise §4.1-4.4</li> </ul>
7	Standard heats of reaction, formation and combustion, Standard heats cont'd, Temperature dependence of Ho, and Heat effects of industrial reactions (Ch. 4)	<ul style="list-style-type: none"> <li>• Upload BPN: §4.5-4.8</li> <li>• Online Quiz §4.5-4.8</li> <li>• Upload Exercise §4.5-4.7</li> </ul>
8	Midterm Exam or Project Deliverable	<ul style="list-style-type: none"> <li>• Possible assignment upload</li> </ul>
9	2nd law of thermo., heat engines, thermodynamic temperature scales (Ch. 5)	<ul style="list-style-type: none"> <li>• Upload BPN: §5.1-5.4</li> <li>• Online Quiz §5.1-5.4</li> <li>• Upload Exercise §5.1-5.4</li> </ul>
10	Mathematical statement of the 2nd law, entropy balance for open systems and the 3rd law of Thermodynamics (Ch. 5)	<ul style="list-style-type: none"> <li>• Upload BPN: §5.5-5.7</li> <li>• Online Quiz §5.5-5.7</li> <li>• Upload Exercise §5.5-5.7</li> </ul>
11	2nd law of thermo., heat engines, thermodynamic temperature scales (Ch. 5) <b>Project 2 Second Law Analysis of Power Cycle Performance</b>	<ul style="list-style-type: none"> <li>• Upload BPN: §5.8-5.11</li> <li>• Online Quiz §5.8-5.11</li> <li>• Upload Exercise §5.8-5.9</li> </ul>
12	Fundamental property relations, residual properties, and generalized correlations for gases PL6	<ul style="list-style-type: none"> <li>• Upload BPN: §6.1-6.4</li> <li>• Online Quiz §6.1-6.4</li> <li>• Upload Exercise §6.1-6.4</li> </ul>
13	Two-phase liquid/vapor systems, thermodynamic diagrams: Mollier diagram Tables of thermodynamic properties (Ch. 6), liquefaction	<ul style="list-style-type: none"> <li>• Upload BPN: §6.5-6.8</li> <li>• Online Quiz §6.5-6.8</li> <li>• Upload Exercise 6.5-6.7</li> </ul>
14	Steam power plants, internal combustion engines, and jet engines	<ul style="list-style-type: none"> <li>• Upload BPN: Ch. 8</li> <li>• Online Quiz on Ch. 8</li> <li>• Upload Exercise in Ch. 8</li> </ul>
15	Carnot refrigerator, vapor-compression cycles, choice of refrigerant, heat pumps <b>Project 3 Design of Refrigeration Cycle Using 1<sup>st</sup> &amp; 2<sup>nd</sup> Laws</b>	<ul style="list-style-type: none"> <li>• Upload BPN: §9.1-9.3</li> <li>• Online Quiz §9.1-9.3</li> <li>• Upload Exercise §9.1-9.3</li> </ul>
16	Project Work Completion <b>Final exam on date as per official schedule</b>	<i>Check official schedule for final exam date</i>

\*This schedule represents a tentative schedule only and is subject to change at the instructor's discretion.

\*\*PL means playlist

## PL1

View screencasts (required):

- (1) Thermodynamics Introduction (9 min, by LearnChemE) <https://www.youtube.com/watch?v=UMs9GlrY4dw>
- (2) Units, Dimension, and Unit Conversions (10 min, by Dr. Patton-Luks) <https://www.youtube.com/watch?v=UY3yE5TTW3w>
- (3) Chapter 1a Terminology (15 min, by Dr. Patton-Luks) <https://www.youtube.com/watch?v=ibOLrBdoXN8&t=337s>
- (4) Chapter 1b Temperature and Pressure (8 min, by Dr. Patton-Luks) <https://www.youtube.com/watch?v=TPK4Okz73NI>
- (5) Chapter 1c Introduction to Energy, Work and Heat (13 min, by Dr. Patton-Luks) <https://www.youtube.com/watch?v=goUVEIn0i60>
- (6) Mass, Moles and Volume (15 min, by Dr. Patton-Luks) <https://www.youtube.com/watch?v=qZtALZL0-v0>
- (7) Compositions (13 min, by Dr. Patton-Luks) [https://www.youtube.com/watch?v=hWxaXe\\_r0](https://www.youtube.com/watch?v=hWxaXe_r0)

View screencasts (supplemental):

- (8) Temperature (5 min, by Dr. Patton-Luks) <https://www.youtube.com/watch?v=GI3tyLGdEd4>
- (9) Pressure (6 min, by Dr. Patton-Luks) <https://www.youtube.com/watch?v=GI3tyLGdEd4>
- (10) Forms of Energy (8 min, by Dr. Patton-Luks) <https://www.youtube.com/watch?v=Es vcpU1XLwO>
- (11) Forms of Energy – Continued (15 min, by Dr. Patton-Luks) <https://www.youtube.com/watch?v=rOGsm7B09z4>

## PL2A

View screencasts, by LearnChemE.com:

- (1) Internal Energy Introduction (3 min) <https://www.youtube.com/watch?v=UTj9YEvOt2U>
- (2) What is Enthalpy? (5 min) <https://www.youtube.com/watch?v=AK9JIFvzqMY>
- (3) Introduction to First Law: Closed System (6min) [https://www.youtube.com/watch?v=-OHRmEaA1TI&list=PL4xAk5aenUiy5I6OsjJ\\_3rdKoO0q04I](https://www.youtube.com/watch?v=-OHRmEaA1TI&list=PL4xAk5aenUiy5I6OsjJ_3rdKoO0q04I)
- (4) First Law Applied to Closed Systems (6 min) <https://www.youtube.com/watch?v=-OHRmEaA1TI>
- (5) Enthalpy Dependence on Pressure (4 min) <https://www.youtube.com/watch?v=O6TBDJGKvWE>
- (6) Work and Enthalpy Misunderstanding (3 min) <https://www.youtube.com/watch?v= pd9nUtLDdU>
- (7) State Functions Explained (5 min) [https://www.youtube.com/watch?v=JXEn\\_AaC63k](https://www.youtube.com/watch?v=JXEn_AaC63k)

## PL2B

View screencasts, by Dr. Patton-Luks:

- (8) Reversible and Irreversible Processes (8 min, by LearnChemE) <https://www.youtube.com/watch?v=yBUgRGYGBcCO>
- (9) Reversible Work in Closed Systems (16 min, by Dr. Patton-Luks) [https://www.youtube.com/watch?v=ofImNRTlaKY&list=PL8tOln8mesDv013XCIAC\\_xs0rVz9fMs35&index=1](https://www.youtube.com/watch?v=ofImNRTlaKY&list=PL8tOln8mesDv013XCIAC_xs0rVz9fMs35&index=1)
- (10) Heat Capacity Definitions (11 min) [https://www.youtube.com/watch?v=O8sONEOboK0&list=PL8tOln8mesDv013XCIAC\\_xs0rVz9fMs35&index=2](https://www.youtube.com/watch?v=O8sONEOboK0&list=PL8tOln8mesDv013XCIAC_xs0rVz9fMs35&index=2)
- (11) Heat Capacity Examples (8 min, by Dr. Patton-Luks) [https://www.youtube.com/watch?v=dvVR-nJW0lk&list=PL8tOln8mesDv013XCIAC\\_xs0rVz9fMs35&index=3](https://www.youtube.com/watch?v=dvVR-nJW0lk&list=PL8tOln8mesDv013XCIAC_xs0rVz9fMs35&index=3)
- (12) Mass & Energy Balance Equations for Open Systems (12 min, by Dr. Patton-Luks) [https://www.youtube.com/watch?v=IuGweyT6m4Y&list=PL8tOln8mesDv013XCIAC\\_xs0rVz9fMs35&index=4](https://www.youtube.com/watch?v=IuGweyT6m4Y&list=PL8tOln8mesDv013XCIAC_xs0rVz9fMs35&index=4)

## PL3A

View screencasts by Dr. Patton-Luks:

- (1) Chapter 3b 1st Law for Closed Systems (11 min) <https://www.youtube.com/watch?v=6zF5F6S8vU8>
- (2) Chapter 3c 1st Law for Steady Flow Systems with Pump Example (11 min) <https://www.youtube.com/watch?v=rToB-ciBdas>
- (3) Chapter 3d Heat Exchanger Example (15 min) <https://www.youtube.com/watch?v=xeV9mZDs qOY>
- (4) Chapter 3e Nozzle Example (19 min) <https://www.youtube.com/watch?v=SoigiaJowuE>
- (5) Chapter 3f Calorimeter Example (8 min) <https://www.youtube.com/watch?v=LwxF9tVSU3Q>
- (6) Chapter 3g Common Steady Flow Processes (11 min) <https://www.youtube.com/watch?v=NVBv05YLpK4>
- (7) Chapter 3h 1st Law for Open Unsteady State Systems (13 min) <https://www.youtube.com/watch?v=rgZhhXkUllw>

## PL3B

View screencasts by Dr. Patton-Luks:

- (1) Ideal Gases and Their Properties (12 min) [https://www.youtube.com/watch?v=OW2bbJFYKCU&list=PL8tOln8mesDv013XCIAC\\_xs0rVz9fMs35&index=5](https://www.youtube.com/watch?v=OW2bbJFYKCU&list=PL8tOln8mesDv013XCIAC_xs0rVz9fMs35&index=5)
- (2) Ideal Gas Special Properties (9 min) [https://www.youtube.com/watch?v=hEIA dUiuxcg&list=PL8tOln8mesDv013XCIAC\\_xs0rVz9fMs35&index=6](https://www.youtube.com/watch?v=hEIA dUiuxcg&list=PL8tOln8mesDv013XCIAC_xs0rVz9fMs35&index=6)
- (3) Ideal Gas Examples (13 min) [https://www.youtube.com/watch?v=V4q10M8tqYE&list=PL8tOln8mesDv013XCIAC\\_xs0rVz9fMs35&index=7](https://www.youtube.com/watch?v=V4q10M8tqYE&list=PL8tOln8mesDv013XCIAC_xs0rVz9fMs35&index=7)

- (4) Cubic Equations of State (EOS) (11 min) [https://www.youtube.com/watch?v=fYwxQBGe-dw&list=PL8tOln8mesDv013XCIAC\\_xs0rVz9fMs35&index=8](https://www.youtube.com/watch?v=fYwxQBGe-dw&list=PL8tOln8mesDv013XCIAC_xs0rVz9fMs35&index=8)
- (5) Generalized Equations for Gases (9 min) [https://www.youtube.com/watch?v=-VW-jcHIOFM&list=PL8tOln8mesDv013XCIAC\\_xs0rVz9fMs35&index=9](https://www.youtube.com/watch?v=-VW-jcHIOFM&list=PL8tOln8mesDv013XCIAC_xs0rVz9fMs35&index=9)
- (6) Pure PVT Method Selection (6 min) [https://www.youtube.com/watch?v=Gv9v6F9yvEE&list=PL8tOln8mesDv013XCIAC\\_xs0rVz9fMs35&index=11](https://www.youtube.com/watch?v=Gv9v6F9yvEE&list=PL8tOln8mesDv013XCIAC_xs0rVz9fMs35&index=11)

View screencasts (supplemental):

- (7) Lecture 3C Introduction to PVT Equations of State (14 min) <https://www.youtube.com/watch?v=q9BaFOcgZ3c&list=PL8tOln8mesDvOaaaGiJvo2yOg4htSzaOY&index=15>
- (8) Lecture 3C Introduction to PVT Equations of State (14 min) <https://www.youtube.com/watch?v=q9BaFOcgZ3c&list=PL8tOln8mesDvOaaaGiJvo2yOg4htSzaOY&index=15>
- (9) Generalized Equations for Liquids (10 min) [https://www.youtube.com/watch?v=wuW0TfFAOvc&list=PL8tOln8mesDv013XCIAC\\_xs0rVz9fMs35&index=10](https://www.youtube.com/watch?v=wuW0TfFAOvc&list=PL8tOln8mesDv013XCIAC_xs0rVz9fMs35&index=10)
- (10) Derivation of Polytropic Work Formula (4 min) <https://www.youtube.com/watch?v=3j2Ybgx0JyE&list=PL8tOln8mesDtRe0CrOAI0SIALDRSVCact&index=2>

#### PL4

View screencasts by Dr. Patton-Luks:

- (1) Thermo 1 Intro to Heats of Reaction (12 min) [https://www.youtube.com/watch?v=it8YcxYcX\\_w](https://www.youtube.com/watch?v=it8YcxYcX_w)

View screencasts by LearnChemE.com:

- (2) Heat of Vaporization: Antoine's Equation (3 min) <https://www.youtube.com/watch?v=uEdgwhPPUE4>
- (3) Heat of Vaporization: Clausius Clapeyron Equation (2 min) <https://www.youtube.com/watch?v=uEdgwhPPUE4>
- (4) Heat of Reaction from Heat of Formation (6 min) <https://www.youtube.com/watch?v=zYA1cYDghUw>

#### PL5

View screencasts by Dr. Patton-Luks:

- (1) The Early Development of the Second Law (11 min) [https://www.youtube.com/watch?v=DQF9pLp\\_WjA&list=PL8tOln8mesDv013XCIAC\\_xs0rVz9fMs35&index=13](https://www.youtube.com/watch?v=DQF9pLp_WjA&list=PL8tOln8mesDv013XCIAC_xs0rVz9fMs35&index=13)
- (2) Entropy Change in Adiabatic Processes - Reversible and Irreversible (4 min) [https://www.youtube.com/watch?v=GAhqFWB5GnM&list=PL8tOln8mesDv013XCIAC\\_xs0rVz9fMs35&index=14](https://www.youtube.com/watch?v=GAhqFWB5GnM&list=PL8tOln8mesDv013XCIAC_xs0rVz9fMs35&index=14)
- (3) Intro to 2<sup>nd</sup> Law for Cycles (9 min) <https://www.youtube.com/watch?v=IZtpxFMqsJO>

View screencasts by LearnChemE.com:

- (4) Heat Engine Introduction (6 min) [https://www.youtube.com/watch?v=log\\_BcG89aA](https://www.youtube.com/watch?v=log_BcG89aA)
- (5) Carnot Heat Engine Calculations (9 min) <https://www.youtube.com/watch?v=m6XaSGOi2wA>
- (6) Entropy: 2 Explanations (3 min) <https://www.youtube.com/watch?v=Ox1A0A6WX3E>
- (7) 2<sup>nd</sup> Law Application - Viability Check (4 min) <https://www.youtube.com/watch?v=tepL-O7jHAQ>
- (8) 2<sup>nd</sup> Law of Thermo - Hilsch Tube Viability Check (8 min) <https://www.youtube.com/watch?v=o4qwEgj9k1A>

#### PL6

View screencasts:

- (1) Lecture 02c Using Partial Derivatives to Evaluate Thermodynamic Properties (11 min) <https://www.youtube.com/watch?v=nLsVXPvS0w>
- (2) Lecture Thermodynamic Property Calculations for Real Substances - Part 1 (20 min) <https://www.youtube.com/watch?v=nLsVXPvS0w>
- (3) Lecture Thermodynamic Property Calculations for Real Substances - Part 2 (7 min) <https://www.youtube.com/watch?v=nLsVXPvS0w>

#### PL8

View screencasts by Dr. Patton-Luks:

- (1) Lecture 02a Liquefaction (5 min) <https://www.youtube.com/watch?v=O9kkNEYbLT8>
- (2) Lecture 02b Power Cycles (6 min) <https://www.youtube.com/watch?v=uEBoKQZgMV0&list=PL8tOln8mesDvOaaaGiJvo2yOg4htSzaOY&index=9>

#### PL9

View screencasts by Dr. Patton-Luks:

- (1) Intro to Refrigeration (10 min) <https://www.youtube.com/watch?v=04Mxwn9jli4>
- (2) Vapor-Compression Cycle in Refrigeration (12 min) <https://www.youtube.com/watch?v=GbprWnGN0tY>
- (3) Intro to Refrigeration Example (10 min) <https://www.youtube.com/watch?v=04Mxwn9jli4>
- (4) Selecting Refrigerants (8 min) <https://www.youtube.com/watch?v=cJ3IGdp9jI>

## **Student Support and Success**

### **John B. Coleman Library**

The John B. Coleman Library's mission is to enhance the scholarly pursuit of knowledge, to foster intellectual curiosity, and to promote life-long learning and research through our innovative services, resources, and cultural programs, which support the Prairie View A&M University's global mission of teaching, service, and research. It maintains library collections and access both on campus, online, and through local agreements to further the educational goals of students and faculty. Website: <https://www.pvamu.edu/library/>; Phone: 936-261-1500

### **Academic Advising Services**

Academic Advising Services offers students a variety of services that contributes to student success and leads towards graduation. We assist students with understanding university policies and procedures that affect academic progress. We support the early alert program to help students get connected to success early in the semester. We help refer students to the appropriate academic support services when they are unsure of the best resource for their needs. Faculty advisors support some students in their respective colleges. Your faculty advisor can be identified in PantherTracks. Advisors with Academic Advising Services are available to all students. We are located across campus. Find your advisor's location by academic major at [www.pvamu.edu/advising](http://www.pvamu.edu/advising). Phone: 936-261-5911

### **The University Tutoring Center**

The University Tutoring Center (UTC) offers free tutoring and academic support to all registered PVAMU students. The mission of the UTC is to help provide a solid academic foundation that enables students to become confident, capable, independent learners. Competent and caring staff and peer tutors guide students in identifying, acquiring, and enhancing the knowledge, skills, and attitudes needed to reach their desired goals. Tutoring and academic support are offered face-to-face in the UTC, in virtual face-to-face sessions (<https://www.pvamu.edu/student-success/sass/university-tutoring-center/>), and through online sessions (<https://www.pvamu.edu/pvplace/>). Other support services available for students include Supplemental Instruction, Study Break, Academic Success Workshops, and Algebra Study Jam. Location: J. B. Coleman Library, Rm. 307; Phone: 936-261-1561; Email: [pv tutoring@pvamu.edu](mailto:pv tutoring@pvamu.edu); Website: <https://www.pvamu.edu/student-success/sass/university-tutoring-center/>

### **Writing Center**

The Writing Center provides well-trained peer tutors to assist students with writing assignments at any stage of the writing process. Tutors help students with various writing tasks from understanding assignments, brainstorming, drafting, revising, editing, researching, and integrating sources. Students have free access to Grammarly online writing assistance. Grammarly is an automated proofreading and plagiarism detection tool. Students must register for Grammarly by using their student email address. In addition, students have access to face-to-face and virtual tutoring services either asynchronously via email or synchronously via Zoom. Location: J. B. Coleman Library, Rm. 209; Phone: 936-261-3724; Website: <https://www.pvamu.edu/student-success/writing-center/>; Grammarly Registration: <https://www.grammarly.com/enterprise/signup>

### **Academic Early Alert**

Academic Early Alert is a proactive system of communication and collaboration between faculty, academic advisors, and PVAMU students that is designed to support student success by promptly identifying issues and allowing for intervention. Academic Early Alerts help students by providing a central location to schedule advising appointments, view advisor contact information, and request assistance. Students who recognize that they have a problem that is negatively affecting their academic performance or ability to continue school may self-refer an Academic Early Alert. To do so, students will log in to PV Place and click on Academic Early Alert on the left sidebar. Phone: 936-261-5902; Website: <https://www.pvamu.edu/student-success/early-alert/>

### **Student Counseling Services**

The Student Counseling Services unit offers a range of services and programs to assist students in maximizing their potential for success: short-term individual, couples, and group counseling, as well as crisis intervention, outreach, consultation, and referral services. The staff is licensed by the State of Texas and assists students who are dealing with academic skills concerns, situational crises, adjustment problems, and emotional difficulties. Information shared with the staff is treated confidentially and in accordance with Texas State Law. Location: Hobart Taylor, 2<sup>nd</sup> floor; Phone: 936-261-3564; Website: <https://www.pvamu.edu/healthservices/student-counseling-services/>

### **Office of Testing Services**

Testing Services serves to create opportunities by offering a suite of exams that aid in the students' academic and professional success. Currently, we administer entrance (HESI A2), college readiness (TSI assessment), Prior Learning (CLEP, DSST), and proctored exams. Location: Wilhelmina Delco, 3<sup>rd</sup> Floor, Rm. 305; Phone: 936-261-3627; Email: [aetesting@pvamu.edu](mailto:aetesting@pvamu.edu); Website: [www.pvamu.edu/testing](http://www.pvamu.edu/testing)

### **Office of Diagnostic Testing and Disability Services**

The Americans with Disabilities Act (ADA) is a federal anti-discrimination statute that provides comprehensive civil rights protection for persons with disabilities. Among other things, this legislation requires that all students with disabilities be guaranteed a learning environment that provides for reasonable accommodation of their disabilities. If you believe you have a disability requiring an accommodation, contact the Office of Disability Services. As a federally-mandated educational support unit, the Office of Disability Services serves as the repository for confidential disability files for faculty, staff, and students. For persons with a disability, the Office develops individualized ADA letters of request for accommodations. Other services include learning style inventories, awareness workshops, accessibility pathways, webinars, computer laboratory with adapted hard and software, adapted furniture, proctoring non-standardized test administrations, ASL interpreters, ALDs, digital recorders, Livescribe, and a comprehensive referral network across campus and the broader community. Location: Hobart Taylor, Rm. 1D128; Phone: 936-261-3583; Website: <https://www.pvamu.edu/disabilityservices/>

### **Center for Instructional Innovation and Technology Services (CIITS)**

Distance Learning, also referred to as Distance Education, is the employment of alternative instructional delivery methods to extend programs and services to persons unable to attend college in the traditional manner. The Center for Instructional Innovation and Technology Services (CIITS) supports student learning through online, hybrid, web-assist, and 2-way video course delivery. For more details and contact information, visit: <https://www.pvamu.edu/dlearning/distance-learning-2-2/students-2/>; Phone: 936-261-3283

### **Veteran Affairs**

Veterans Services works with student veterans, current military and military dependents to support their transition to the college environment and continued persistence to graduation. The Office coordinates and certifies benefits for both the G.I. Bill and the Texas Hazlewood Act. Location: Evans Hall, Rm. 102; Phone: 936-261-3563; Website: <https://www.pvamu.edu/sa/departments/veteranaffairs/>

### **Office for Student Engagement**

The Office for Student Engagement delivers comprehensive programs and services designed to meet the co-curricular needs of students. The Office implements inclusive and accessible programs and services that enhance student development through exposure to and participation in diverse and relevant social, cultural, intellectual, recreational, community service, leadership development, and campus governance. Location: Memorial Student Center, Rm. 221; Phone: 936-261-1340; Website: <https://www.pvamu.edu/studentengagement/>

### **Career Services**

Career Services supports students through professional development, career readiness, and placement and employment assistance. The Office provides one-on-one career coaching, interview preparation, resume and letter writing, and career exploration workshops and seminars. Services are provided for students at the Northwest Houston Center and College of Nursing in the Medical Center twice a month or on a requested basis. Distance Learning students are encouraged to visit the Career Services website for information regarding services provided. Location: Anderson Hall, 2<sup>nd</sup> floor; Phone: 936-261-3570; Website: <https://www.pvamu.edu/careerservices/>

## **University Rules and Procedures**

### **Academic Misconduct**

Academic dishonesty is defined as any form of cheating or dishonesty that has the effect or intent of interfering with any academic exercise or fair evaluation of a student's performance. The college faculty can provide additional information, particularly related to a specific course, laboratory, or assignment.

You are expected to practice academic honesty in every aspect of this course and all other courses. Make sure you are familiar with the *University Administrative Guidelines on Academic Integrity*, which can be found on the [Academic Integrity webpage](#). Students who engage in academic misconduct are subject to university disciplinary procedures. As listed in the *University Administrative Guidelines on Academic Integrity*, the University Online Catalog, and the Student Code of Conduct, the following are examples of prohibited conduct. This list is not designed to be all-inclusive

or exhaustive. In addition to academic sanctions, any student found to have committed academic misconduct that is also a violation of criminal law may also be subject to disciplinary review and action by the Office of Student Conduct (as outlined in the Student Code of Conduct).

### **Forms of Academic Dishonesty:**

1. **Cheating:** Deception in which a student misrepresents that he/she has mastered information on an academic exercise that he/she has not learned, giving or receiving aid unauthorized by the instructor on assignments or examinations. Examples: unauthorized use of notes for a test; using a "cheat sheet" on a quiz or exam; any alteration made on a graded test or exam which is then resubmitted to the teacher;
2. **Plagiarism:** Careless or deliberate use of the work or the ideas of another; representation of another's work, words, ideas, or data as your own without permission or appropriate acknowledgment. Examples: copying another's paper or answers, failure to identify information or essays from the internet and submitting or representing it as your own; submitting an assignment which has been partially or wholly done by another and claiming it as yours; not properly acknowledging a source which has been summarized or paraphrased in your work; failure to acknowledge the use of another's words with quotation marks;
3. **Collusion:** When more than one student or person contributes to a piece of work that is submitted as the work of an individual;
4. **Conspiracy:** Agreeing with one or more persons to commit an act of academic/scholastic dishonesty; and
5. **Multiple Submission:** Submission of work from one course to satisfy a requirement in another course without explicit permission. Example: using a paper prepared and graded for credit in one course to fulfill a requirement and receive credit in a different course.

### **Nonacademic Misconduct**

The university respects the rights of instructors to teach and students to learn. Maintenance of these rights requires campus conditions that do not impede their exercise. Campus behavior that interferes with either (1) the instructor's ability to conduct the class, (2) the inability of other students to profit from the instructional program, or (3) campus behavior that interferes with the rights of others will not be tolerated. An individual engaging in such disruptive behavior may be subject to disciplinary action. The Office of Student Conduct will adjudicate such incidents under nonacademic procedures.

### **Sexual Misconduct**

Sexual harassment of students and employees at Prairie View A&M University is unacceptable and will not be tolerated. Any member of the university community violating the university's sexual harassment policy will be subject to disciplinary action. In accordance with the Texas A&M University System guidelines, your instructor is obligated to report to the Office of Title IX Compliance ([titleixteam@pvamu.edu](mailto:titleixteam@pvamu.edu)) any instance of sexual misconduct involving a student, which includes sexual assault, stalking, dating violence, domestic violence, and sexual harassment, about which the instructor becomes aware during this course through writing, discussion, or personal disclosure. The faculty and staff of PVAMU actively strive to provide a learning, working, and living environment that promotes respect that is free from sexual misconduct, discrimination, and all forms of violence. If students, faculty, or staff would like assistance or have questions, they may contact the Title IX Coordinator at 936-261-2144 or [titleixteam@pvamu.edu](mailto:titleixteam@pvamu.edu). More information can be found at [www.pvamu.edu/titleix](http://www.pvamu.edu/titleix), including confidential resources available on campus.

### **Pregnancy, Pregnancy-related, and Parenting Accommodations**

Title IX of the Education Amendments of 1972 prohibits sex discrimination, which includes discrimination based on pregnancy, marital status, or parental status. Students seeking accommodations related to pregnancy, pregnancy-related conditions, or parenting (reasonably immediate postpartum period) are encouraged to contact Student Disability Services or the Dean of Students' Office for additional information and to request accommodations.

### **Non-Discrimination Statement**

Prairie View A&M University does not discriminate on the basis of race, color, sex, religion, national origin, age, disability, genetic information, veteran status, sexual orientation, or gender identity in its programs and activities. The University is committed to supporting students and complying with The Texas A&M University System non-discrimination policy. It seeks to establish an environment that is free of bias, discrimination, and harassment. If you

experience an incident of discrimination or harassment, we encourage you to report it. If you would like to speak with someone who may be able to afford you privacy or confidentiality, there are individuals who can meet with you. The Director of Equal Opportunity & Diversity has been designated to handle inquiries regarding the non-discrimination policies and can be reached at Harrington Science Building, Suite 109 or by phone at 936-261-1744 or 1792.

### **Class Attendance Policy (See the University Online Catalog for Full Attendance Policy)**

Prairie View A&M University requires regular class attendance. Attending all classes supports the full academic development of each learner, whether classes are taught with the instructor physically present or via distance learning technologies such as interactive video and/or the internet. Excessive absenteeism, whether excused or unexcused, may result in a student's course grade being reduced or in the assignment of a grade of "F." Absences are accumulated beginning with the first day of class during regular semesters and summer terms. Each faculty member will include the University's attendance policy in each course syllabus.

### **Student Academic Appeals Process**

Authority and responsibility for assigning grades to students rest with the faculty. However, in those instances where students believe that miscommunication, errors, or unfairness of any kind may have adversely affected the instructor's assessment of their academic performance, the student has a right to appeal by the procedure listed in the University Online Catalog and by doing so within thirty days of receiving the grade or experiencing any other problematic academic event that prompted the complaint.

### **Technical Considerations**

#### **Minimum Recommended Hardware and Software:**

- Intel PC or Laptop with Windows 10 or later version; Mac with OS High Sierra\*
- Smartphone or iPad/Tablet with Wi-Fi\*
- High-speed Internet access
- 8 GB Memory
- Hard drive with 320 GB storage space
- 15" monitor, 800x600, color or 16 bit
- Sound card w/speakers
- Microphone and recording software
- Keyboard & mouse
- Most current version of Google Chrome, Safari, or Firefox

**Note:** Be sure to enable Java & pop-ups in the Web browser preferences

\* Smartphones, Google Chrome books, and Android tablets may not be supported. iPads are the only tablets supported.

#### **Participants should have a basic proficiency of the following computer skills:**

- Sending and receiving email
- A working knowledge of the Internet
- Microsoft Word (or a program convertible to Word)
- Acrobat PDF Reader
- Windows or Mac OS
- Video conferencing software

#### **Netiquette (online etiquette)**

Students are expected to participate in all discussions and virtual classroom chats as directed. Students are to be respectful and courteous to others on discussion boards. Foul or abusive language will not be tolerated. Do not use ALL CAPS for communicating to others AS IT CAN BE INTERPRETED AS YELLING. Avoid slang terms such as "wassup?" and texting abbreviations such as "u" instead of "you." Limit and possibly avoid the use of emoticons. Be cautious when using humor or sarcasm as tone is sometimes lost in an email or discussion post, and the message might be taken seriously or sound offensive.

#### **Video Conferencing Etiquette**

When using Zoom, WebEx, or other video conferencing tools, confirm the visible area is tidy, clear of background clutter, inappropriate or offensive posters, and other distractions. Ensure you dress appropriately and avoid using

high traffic or noisy areas. Stay muted when you are not speaking and avoid eating/drinking during the session. Before the class session begins, test audio, video, and lighting to alleviate technology issues.

### **Technical Support**

Students should go to <https://mypassword.pvamu.edu/> if they have password issues. The page will provide instructions for resetting passwords and contact information if login issues persist. For other technical questions regarding eCourses, call the Center for Instructional Innovation and Technology Services at 936-261-3283 or email [ciits@pvamu.edu](mailto:ciits@pvamu.edu).

### **Communication Expectations and Standards**

Emails or discussion postings will receive a response from the instructor, usually in less than 48 hours. Urgent emails should be marked as such. Check regularly for responses.

### **Discussion Requirement**

Online courses often require minimal to no face-to-face meetings. However, conversations about the readings, lectures, materials, and other aspects of the course can occur in a seminar fashion. The use of the discussion board will accomplish this. The instructor will determine the exact use of discussion boards.

**It is strongly suggested** that students type their discussion postings in a word processing application such as Word and save it to their PC or a removable drive before posting to the discussion board. This is important for two reasons: 1) If for some reason your discussion responses are lost in your online course, you will have another copy; 2) Grammatical errors can be greatly minimized by the use of the spell-and-grammar check functions in word processing applications. Once the post(s) have been typed and corrected in the word processing application, copy and paste to the discussion board.

### **COVID-19 Campus Safety Measures [NOTE: Delete this section when the COVID-19 pandemic is over]**

To promote public safety and protect students, faculty, and staff during the coronavirus pandemic, PVAMU has adopted policies and practices to limit virus transmission.

- **Self-monitoring** - Students should follow CDC recommendations for self-monitoring. Students who have a fever or exhibit symptoms of COVID-19 should participate in class remotely and should not participate in face-to-face instruction.
- **Face Coverings** - Face coverings (cloth face covering, surgical mask, etc.) are recommended in classrooms, teaching laboratories, common spaces such as lobbies and hallways, public study spaces, libraries, academic resource, and support offices, and outdoor spaces where 6 feet of physical distancing is challenging to maintain reliably.
- **Physical Distancing** - Physical distancing should be maintained between students, instructors, and others in course and course-related activities where possible.
- **Personal Illness and Quarantine** - Students required to quarantine are to participate in courses and course-related activities remotely and must not attend face-to-face course activities. Students should notify their instructors of the quarantine requirement. Students under quarantine are expected to participate in courses and complete graded work unless they have symptoms that are too severe to participate in course activities. Students experiencing personal injury or illness that is too severe for the student to attend class qualify for an excused absence. To receive an excused absence, students must provide appropriate documentation to the Office for Student Conduct, [studentconduct@pvamu.edu](mailto:studentconduct@pvamu.edu).